

A FERN CONSTITUENT, FERNENE

A TRITERPENOID HYDROCARBON OF A NEW TYPE

Hiroyuki Ageta, Kenji Iwata and Shinsaku Natori

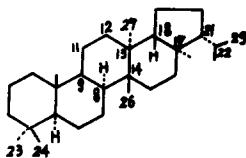
Showa College of Pharmacy, Setagaya-ku, Tokyo, Japan

(Received 3 July 1963)

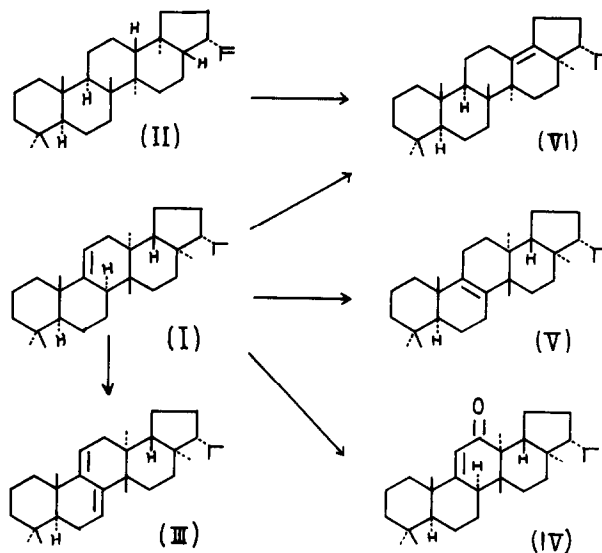
IN a previous communication<sup>(1)</sup> we had reported the isolation of two triterpenoid hydrocarbons, fernene,  $C_{30}H_{50}$ , m.p. 170-171°,  $[\alpha]_D^{16} -16.5$  ( $c=0.79$ ,  $CHCl_3$ ), one olefinic proton at  $\delta^{CCl_4} 5.28$  ppm., and diploptene (hopene-b, II) from the leaflets of a fern, Dryopteris crassirhizoma NAKAI (Aspidiaceae). Now it has been shown that fernene has structure I.

Oxidation of fernene with selenium dioxide in a solution of acetic acid gave ferna-7,9(11)-diene (III)\*,  $C_{30}H_{48}$ , m.p. 198-199°,  $[\alpha]_D^{20} -157.3$  ( $c=0.50$ ,  $CHCl_3$ ), having  $\lambda_{max}^{EtOH} 232, 240, 248$   $\mu$  ( $\epsilon 10,200, 10,800, 6,600$ ), characteristic of a conjugated heteroannular diene. Chromic acid oxidation of fernene in

\*The name, fernaene is proposed for the following saturated hydrocarbon corresponding to fernene.



acetic acid afforded 12-keto-fern-9(11)-ene (IV),  $C_{30}H_{48}O$ , m.p. 221.5-223°,  $[\alpha]_D^{17} -28.0$  ( $c=1.00$ ,  $CHCl_3$ ),  $\lambda_{max}^{EtOH} 246 m\mu$  ( $\epsilon 9,000$ ),  $\nu_{max}^{KBr} 1672, 1611 cm^{-1}$ , which showed a positive Cotton effect of molecular amplitude ( $a$ ) +12,000 in the optical rotatory dispersion curve<sup>(3)</sup>.



The mass spectra of fernene, isofernene (V)<sup>(1)</sup>, m.p. 184-185°,  $[\alpha]_D^{16} +15.2^\circ$  ( $CHCl_3$ ), and fern-7,9(11)-diene (III) showed characteristic base peaks at  $m/e$  243, 243 and 255, respectively. Placement of methyl groups at C-13 and C-14, and double bonds at 9(11) (in fernene) and 7,9(11) (in fernadiene) follows from the fact that the same base peaks at  $m/e$  243 and 255 were observed in arborene ( $\Delta^{9(11)}$ ) and arboradiene ( $\Delta^{7,9(11)}$ )<sup>(2)</sup>, and the identity of fernene with davallene<sup>(3, 4)</sup>. The configurations of the two methyl groups

were concluded to be 13 $\alpha$  and 14 $\beta$  on the basis of similarities of the UV spectrum of fernene-7,9(11)-diene with those of corresponding dienes from euphol<sup>(5)</sup>, multiflorenol<sup>(6)</sup> and bauerenol<sup>(7)</sup>. This was also confirmed by the highly negative change of  $[\alpha]_D$  values from isofernene (V) to the diene III, and the O.R.D. curve<sup>(2, 3)</sup> of 12-keto-fernene-9(11)-ene (IV).

Treatment of fernene with 6% hydrochloric acid in boiling acetic acid for 20 hours yielded a hydrocarbon (m.p. 196.5-198°) in addition to isofernene. It was found that this hydrocarbon (VI) was identical with hopene-II<sup>(8,9)</sup> by mixed melting point, infrared spectra, and gas chromatography retention time using SE-30 and QF-1 columns.

Biogenetically, fernene is assumed to be formed from diploptene by the migration of hydrogen and methyl groups (21 $\beta$ H $\rightarrow$ 22, 17 $\beta$ H $\rightarrow$ 21 $\beta$ , 18 $\alpha$ CH<sub>3</sub> $\rightarrow$ 17 $\alpha$ , 13 $\beta$ H $\rightarrow$ 18 $\beta$ , 14 $\alpha$ CH<sub>3</sub> $\rightarrow$ 13 $\alpha$ , 8 $\beta$ CH<sub>3</sub> $\rightarrow$ 14 $\beta$ , and 9 $\alpha$ H $\rightarrow$ 8 $\alpha$ ), followed by formation of the 9(11)-double bond. The configuration at carbons 17 and 21 in fernenes and hopene-II were assigned on the same basis.

Isolation of the onocerane<sup>(10, 11)</sup> and zeorinane (hopane) series<sup>(1,9)</sup> have already been reported, and thus fernene is the third triterpene skeleton from fern plants.

We wish to thank Prof. K. Nakanishi, Tohoku University, for his kind encouragements and helpful discussion, and also to thank Prof. C. Djerassi and Dr. H. Vorbrüggen, Stanford Univ., for determinations of mass spectra of fernenes and related compounds, and for a manuscript of their paper on arborinol before publication. Thanks are due to Dr. N. Ikekawa, Institute for physical and Chemical Research, for gas chromatography, and to the Takeda Research Laboratory

for n.m.r. spectra determination.

REFERENCES

- 1 H. Ageta, K. Iwata and K. Yonezawa, Chem. Pharm. Bull. (Tokyo) 11, 408 (1963).
- 2 H. Vorbrüggen, S. C. Pakrashi and C. Djerassi, Ann., in press.
- 3 K. Nakanishi, Y. Y. Lin, H. Kakisawa, H. Y. Hsü and H. C. Hsiu, following paper.
- 4 The identity was disclosed by Dr. Vorbrueggen at Stanford.
- 5 G. Ourisson. P. Crabbé, Les triterpène tétracycliques, p. 154, Hermann, Paris (1961).
- 6 P. Sengupta and H. N. Khastgir, Tetrahedron 19, 123 (1963).
- 7 F. N. Lahey and M. V. Leeding, Proc. Chem. Soc. (London) 1958, 342.
- 8 H. Fazakerley, T. G. Halsall and E. R. H. Jones, J. Chem. Soc. 1959, 1877.
- 9 H. Ageta, K. Iwata and Y. Ootake, Chem. Pharm. Bull. (Tokyo) 11, 407 (1963).
- 10 Y. Inubushi, Y. Tsuda and T. Sano, Yakugaku Zasshi 82, 1083 (1962).
- 11 H. Ageta, K. Iwata and Y. Ootake, Chem. Pharm. Bull. (Tokyo) 10, 637 (1962).